

Periodic Table of the Elements



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Frequently used fundamental physical constants

1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³Cs

speed of light in vacuum	c	299 792 458 m s ⁻¹ (<i>exact</i>)
Planck constant	h	6.6261 x 10 ⁻³⁴ J s (<i>ħ</i> =h/2π)
elementary charge	e	1.6022 x 10 ⁻¹⁹ C
electron mass	m_e	9.1094 x 10 ⁻³¹ kg
proton mass	m_p	1.6726 x 10 ⁻²⁷ kg
fine-structure constant	α	1/137.036
Rydberg constant	R_∞	10 973 732 m ⁻¹
Boltzmann constant	k	1.3807 x 10 ⁻²³ J K ⁻¹

† Based upon ¹²C. () Indicates the mass number of the most stable isotope.

Alkali Metals	Post-Transition Metals
Alkaline Earth Metals	Metalloids
Lanthanoids	Other-Non Metals
Actinoids	Halogens
Transition Metals	Noble Gases

Solids	Gases
Liquids	Artificial

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18

s-block

Atomic Number → 6

Name → Carbon

Symbol → **C**

Atomic Weight → 12.0107

Ground-State Configuration → [He]2s²2p²

Ionization Energy (eV) → 11.2603

d-block

57 ² D _{3/2} Lanthanum La 138.9055 [Xe]5d ¹ 6s ² 5.5769	58 ¹ G ₄ Cerium Ce 140.116 [Xe]4f ¹ 5d ¹ 6s ² 5.5387	59 ⁴ F _{9/2} Praseodymium Pr 140.90765 [Xe]4f ³ 6s ² 5.473	60 ³ I ₄ Neodymium Nd 144.24 [Xe]4f ⁴ 6s ² 5.5250	61 ¹ H _{5/2} Promethium Pm [145] [Xe]4f ⁵ 6s ² 5.582	62 ⁷ F ₀ Samarium Sm 150.36 [Xe]4f ⁶ 6s ² 5.6437	63 ⁸ S _{7/2} Europium Eu 151.964 [Xe]4f ⁷ 6s ² 5.6704	64 ⁹ D ₂ Gadolinium Gd 157.25 [Xe]4f ⁷ 5d ¹ 6s ² 6.1498	65 ⁶ H _{15/2} Terbium Tb 158.92534 [Xe]4f ⁹ 6s ² 5.8638	66 ⁵ I ₃ Dysprosium Dy 162.500 [Xe]4f ¹⁰ 6s ² 5.9389	67 ⁴ I _{15/2} Holmium Ho 164.93032 [Xe]4f ¹¹ 6s ² 6.0215	68 ³ H ₆ Erbium Er 167.259 [Xe]4f ¹² 6s ² 6.1077	69 ² F _{7/2} Thulium Tm 168.93421 [Xe]4f ¹³ 6s ² 6.1843	70 ¹ S ₀ Ytterbium Yb 173.04 [Xe]4f ¹⁴ 6s ² 6.2542	71 ² D _{3/2} Lutetium Lu 174.967 [Xe]4f ¹⁴ 5d ¹ 6s ² 5.4259
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f-block

89 ² D _{3/2} Actinium Ac [227] [Rn]6d ¹ 7s ² 5.17	90 ³ F ₂ Thorium Th 232.0381 [Rn]6d ² 7s ² 6.3067	91 ⁴ K _{11/2} Protactinium Pa 231.03588 [Rn]5f ² 6d ¹ 7s ² 5.89	92 ⁵ L ₆ Uranium U 238.02891 [Rn]5f ³ 6d ¹ 7s ² 6.1941	93 ⁶ L _{11/2} Neptunium Np [237] [Rn]5f ⁴ 6d ¹ 7s ² 6.2657	94 ⁷ F ₀ Plutonium Pu [244] [Rn]5f ⁶ 7s ² 6.0260	95 ⁸ S _{7/2} Americium Am [243] [Rn]5f ⁷ 7s ² 5.9738	96 ⁹ D ₂ Curium Cm [247] [Rn]5f ⁸ 6d ¹ 7s ² 5.9914	97 ⁶ H _{15/2} Berkelium Bk [247] [Rn]5f ⁹ 7s ² 6.1979	98 ⁵ I ₃ Californium Cf [251] [Rn]5f ¹⁰ 7s ² 6.2817	99 ⁴ I _{15/2} Einsteinium Es [252] [Rn]5f ¹¹ 7s ² 6.42	100 ³ H ₆ Fermium Fm [257] [Rn]5f ¹² 7s ² 6.50	101 ² F _{7/2} Mendelevium Md [258] [Rn]5f ¹³ 7s ² 6.58	102 ¹ S ₀ Nobelium No [259] [Rn]5f ¹⁴ 7s ² 6.65	103 ² P _{1/2} Lawrencium Lr [262] [Rn]5f ¹⁴ 6d ¹ 7s ² 4.9 ?
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